



## THE CRUCIAL ROLE OF MICROORGANISMS IN GLOBAL BIOGEOCHEMICAL CYCLES: MECHANISMS, INTERACTIONS, AND ENVIRONMENTAL IMPLICATIONS

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### ABSTRACT

This study outlines the fundamental role of microorganisms including bacteria, archaea, and fungi as primary drivers of global biogeochemical cycles. Data were collected through a comprehensive literature review of various scientific databases, covering recent publications on the contribution of microorganisms to biogeochemical cycles. It explores how microorganisms facilitate the transformation of essential elements such as carbon, nitrogen, phosphorus, and sulfur through various metabolic pathways, ranging from atmospheric gas fixation to the decomposition of complex organic matter. It also highlights environmental factors that influence microbial activity, as well as the ecological and environmental implications of their roles, including contributions to climate regulation, soil fertility, environmental purification, and the impact of anthropogenic activities. A comprehensive understanding of microbial ecology is crucial for predicting and managing environmental systems amid global challenges such as climate change and pollution.

**Keywords:** Microorganisms, biogeochemical cycles, mechanisms, environmental implications

### ABSTRAK

*Studi ini menguraikan peran fundamental mikroorganisme termasuk bakteri, archaea, dan jamur sebagai penggerak utama siklus biogeokimia global. Data dikumpulkan melalui tinjauan literatur komprehensif dari berbagai basis data ilmiah, mencakup publikasi terkini mengenai kontribusi mikroorganisme dalam siklus biogeokimia. Dibahas secara mendalam bagaimana mikroorganisme memfasilitasi transformasi elemen-elemen esensial seperti karbon, nitrogen, fosfor, dan sulfur melalui berbagai jalur metabolisme, mulai dari fiksasi gas atmosfer hingga dekomposisi bahan organik kompleks. Tulisan ini juga menyoroti faktor-faktor lingkungan yang memengaruhi aktivitas mikroba, serta implikasi ekologis dan lingkungan dari peran mereka, termasuk kontribusi terhadap regulasi iklim, kesuburan tanah, pemurnian lingkungan, dan dampak aktivitas antropogenik. Pemahaman komprehensif tentang ekologi mikroba sangat krusial untuk memprediksi dan mengelola sistem lingkungan di tengah tantangan global seperti perubahan iklim dan polusi.*

**Kata kunci:** Mikroorganisme, siklus biogeokimia, mekanisme, implikasi lingkungan

## 1. Introduction

### 1.1 Definition and Significance of Biogeochemical Cycles

Biogeochemical cycles describe the pathways through which essential chemical elements, such as carbon and nitrogen, are circulated and recycled between biotic (living) and abiotic (non-living) components of ecosystems. Rather than being lost, these elements are stored or cycled through long-term reservoirs known as "sinks". These cycles are essential for sustaining life on Earth, ensuring the availability of nutrients needed for organismal growth and development, and maintaining the dynamic balance of ecosystems as a whole (Dontsova et al, 2020).

A significant aspect of biogeochemical cycles is not merely the transfer of elements among environmental compartments, but more fundamentally, their chemical transformation from one form to another. For instance, atmospheric nitrogen ( $N_2$ ) must be converted into ammonium ( $NH_4^+$ ) to be utilized by plants. Likewise, hydrogen sulfide ( $H_2S$ ) is oxidized into sulfate ( $SO_4$ ), and atmospheric carbon dioxide ( $CO_2$ ) is fixed into organic molecules. If these elements only circulated without undergoing chemical form changes, the bioavailability of these elements would be restricted. Microorganisms are the primary biochemical agents that facilitate these conversions. Their unique ability to change the chemical form of elements makes them accessible to life, positioning them as the planet's "transformers" or "biochemical engineers," rather than merely "transporters" of matter (Singh, 2024).

### 1.2. Microorganisms as Primary Drivers of Biogeochemical Cycles

Despite their microscopic size and often invisibility to the naked eye, microorganisms (microbes) are irreplaceable primary drivers in regulating biogeochemical systems in almost all environments on the planet, from soil and water to extreme environments such as hot springs or deep-sea sediments. The collective metabolic processes they perform, such as nitrogen fixation, carbon fixation, methane metabolism, and sulfur metabolism, directly influence the rate and direction of global biogeochemical cycles. They form the backbone of every ecological system, especially in environments without light where energy cannot be harvested through macroscopic photosynthesis (Schlesinger & Bernhardt, 2020).

Microorganisms are not merely contributors; they are true ecological architects underpinning the function of global ecosystems. The existence and survival of more complex life forms (eukaryotes) fundamentally depend on microbial activity. Some studies even suggest that global biogeochemistry would likely not change drastically even if eukaryotic life were entirely absent. This statement indicates that if eukaryotes disappeared, fundamental elemental cycles would likely continue due to microbial activity. Conversely, if microbes disappeared, complex life would rapidly perish due to a lack of bioavailable nutrients and the accumulation of waste. Even the presence of oxygen in Earth's atmosphere is a result of the photosynthetic activity of ancient microbes. This confirms that without microbes, complex life would not be possible or sustainable, highlighting their role as true architects shaping a habitable planet (Philippot et al, 2021; Dey, 2024).

## 2. Major Groups of Microorganisms and Their Unique Contributions

The metabolic diversity of microorganisms allows them to thrive in various environments and perform a wide range of biogeochemical functions. These adaptations include the ability to survive in extreme conditions such as high temperatures, high salinity, or low pH

### 2.1 Bacteria: Metabolic Diversity and Dominant Role

Bacteria are indispensable microscopic agents, acting as primary drivers in regulating the flow and transformation of essential elements throughout the planet's biogeochemical systems. Their ubiquitous presence ranging from dry soil, vast oceans, lake sediments, to extreme environments like volcanic hot springs or the dark deep seafloor ensures that nutrient cycles continuously operate (Sørensen et al, 2025). Comprehensively, the role of bacteria in biogeochemical cycles can be synthesized as follows:

In the nitrogen cycle, bacteria mediate almost all its transformations. In both terrestrial and aquatic environments, they perform nitrogen fixation (converting atmospheric  $N_2$  into ammonia/ammonium), which is essential for introducing nitrogen into the biosphere, either symbiotically or free-living. Subsequently, in soil and water columns, nitrifying bacteria convert ammonium into nitrate, a form more readily absorbed by most plants. Conversely, under anaerobic conditions, denitrifying bacteria return nitrogen to the atmosphere as gas ( $N_2$  or  $N_2O$ ), closing the cycle. The entire process of ammonification and mineralization of organic nitrogen from dead matter also relies entirely on the activity of decomposer bacteria, which return nitrogen to an inorganic form for recycling (Sanjuan et al, 2020).

In the carbon cycle, bacteria, along with algae, are significant primary producers through carbon fixation. Although macroscopic photosynthesis dominates in light-exposed surfaces, chemoautotrophic bacteria fix carbon dioxide in light-deprived environments, forming the base of food webs in those ecosystems. The most universal role of bacteria is as primary decomposers; in every environment such as soil, fresh water, oceans, bacteria break down dead organic matter, releasing carbon dioxide back into the atmosphere or water through respiration. This process is vital for recycling carbon and other nutrients. In certain anaerobic environments, methanogenic bacteria produce methane ( $CH_4$ ), a potent greenhouse gas, but in other environments, methanotrophic bacteria oxidize this methane, reducing its emissions to the atmosphere (Moran et al, 2022; Bertini & Azevedo, 2022).

For the phosphorus cycle, although it lacks a gaseous phase, bacteria are crucial in regulating its availability in soil and water. Decomposer bacteria facilitate the mineralization of organic phosphorus from dead matter into inorganic phosphate, a form that plants can absorb. Some bacteria also have the ability to solubilize phosphate from insoluble minerals, increasing phosphorus availability for other organisms, while on the other hand, bacteria can also immobilize phosphorus in their biomass, controlling its availability in the environment (Li et al, 2021).

Finally, in the sulfur cycle, bacteria are highly dominant in the transformation of sulfur at various oxidation states. In anaerobic environments, sulfate-reducing bacteria convert sulfate (a relatively available form) into hydrogen sulfide ( $H_2S$ ).

Conversely, in aerobic or anoxic light-exposed environments, sulfur-oxidizing bacteria (both photoautotrophic and chemoautotrophic) convert  $H_2S$  or elemental sulfur into sulfate, completing the cycle. The decomposition process of organic matter containing sulfur is also mediated by bacteria, releasing sulfur back into the environment (Zhang et al, 2022).

In summary, bacteria are biological catalysts that drive the transformation of these key elements through various metabolic pathways. They ensure efficient nutrient recycling, sustain ecosystem productivity, influence atmospheric composition, and ultimately, maintain the conditions that support life on planet Earth. Their adaptive capabilities allow them to perform these vital roles in every environmental niche, from the most fertile to the most extreme.

The extensive metabolic diversity of bacteria fosters functional redundancy within ecosystems, meaning various bacterial groups can perform similar biogeochemical tasks, such as breaking down organic matter. This redundancy is a critical ecological concept that enhances ecosystem resilience, ensuring that essential nutrient cycles and ecosystem processes remain stable even when certain bacterial groups are adversely affected by environmental disturbances. A prime example is denitrification in the nitrogen cycle: the vital conversion of nitrate ( $NO_3^-$ ) to nitrogen gas ( $N_2/N_2O$ ), crucial for preventing eutrophication and for wastewater treatment, is performed by a wide array of taxonomically distinct bacterial genera (e.g., *Pseudomonas*, *Bacillus*, etc.). Despite their varying environmental preferences, these diverse groups collectively ensure that if one is disturbed by operational changes (like pH shifts or toxins), others can take over, thereby maintaining the critical function of nitrate removal and preventing environmental pollution (Ramond, et al, 2025).

## 2.2 Archaea: Contributions in Extreme Environments and Methanogenesis

Archaea are known for their ability to survive and thrive in extreme environments, such as thermophiles in hot springs, halophiles in salt lakes, and acidophiles in acid mine drainage. They are major contributors to methane production (methanogenesis) under anaerobic conditions, as performed by methanogenic archaea groups like *Methanococcus*, *Methanobacteria*, *Methanosarcina*. This process releases methane ( $CH_4$ ), a very potent greenhouse gas, into the atmosphere. Additionally, archaea are also involved in ammonia oxidation (Teske et al, 2021).

Recent research indicates that archaea play a crucial and multifaceted role in the context of climate change, primarily due to their significant involvement in greenhouse gas cycles and their adaptability to continuously changing environmental conditions. As primary methane ( $CH_4$ ) producers in anoxic environments such as sediments and wetlands, methanogenic archaea directly contribute to the increase of this potent greenhouse gas, with their activity potentially accelerated by rising global temperatures. Conversely, other groups of archaea also perform vital anaerobic methane oxidation (AOM) in marine sediments, preventing large quantities of methane from reaching the atmosphere and acting as important natural sinks. Furthermore, archaea are increasingly recognized for their role in the degradation of complex organic matter in large carbon reservoirs like marine sediments and permafrost; this decomposition activity releases carbon as  $CO_2$  or  $CH_4$ , and permafrost thaw due to global warming can activate archaea to release more carbon, creating a positive

feedback loop. In the nitrogen cycle, ammonia-oxidizing archaea (AOA) are major contributors to nitrification in oceans, influencing the crucial availability of nitrogen for marine productivity and CO<sub>2</sub> uptake. Given archaea's adaptability to extreme environments and changing conditions, a deeper understanding of their ecology and physiology, as well as their responses to temperature and oxygen availability, is essential. This is because the role of archaea, often previously overlooked in global climate models, implies that the carbon budget and greenhouse gas cycles may not be fully understood. Therefore, integrating more accurate data on archaeal activity into climate models will enhance their reliability, leading to more realistic predictions of future climate change (Li et al, 2024; Winkel et al, 2018).

### **2.3. Fungi: Primary Decomposers and Symbiotic Associations**

Fungi are critically important decomposers of complex organic matter, especially recalcitrant materials like lignin and cellulose. This decomposition process begins with substrate colonization, followed by the secretion of potent extracellular enzymes such as laccases and peroxidases (for lignin), as well as cellulases and hemicellulases (for cellulose). These enzymes facilitate the depolymerization and hydrolysis of macromolecules into smaller molecules that can be absorbed by fungi as nutrients. Through respiration and internal metabolism, fungi then perform mineralization and nutrient recycling, releasing CO<sub>2</sub> into the atmosphere and returning other inorganic nutrients to the soil, thereby increasing nutrient availability and ecosystem fertility (Berg et al, 2020).

The role of fungi as decomposers of recalcitrant materials complements the role of bacteria, which tend to dominate the decomposition of labile compounds. This indicates an efficient ecological division of labor between different microbial groups, which collectively ensures the thorough decomposition of various types of organic matter. This process often involves microbial succession, where bacteria dominate the initial stages of labile compound decomposition, followed by fungi breaking down more difficult material.

In addition to decomposition, fungi also form crucial mycorrhizal associations with plant roots, significantly aiding in the absorption of nutrients, particularly phosphorus, from the soil. These associations extend the reach of plant roots and enhance access to hard-to-reach nutrients. There are two main types of mycorrhiza that differ in their structure and interaction with plant roots: Endomycorrhiza (specifically Arbuscular Mycorrhiza) and Ectomycorrhiza. Arbuscular Mycorrhiza (AM) is characterized by the fungus penetrating the cortical cells of the plant root, forming internal branching structures called arbuscules for nutrient exchange, and vesicles for storage. This type is very common, found in about 80% of plant species, including most agricultural and herbaceous plants. In contrast, Ectomycorrhiza forms a dense sheath (mantle) around the root tip externally and a network of hyphae (Hartig net) between the cortical cells of the root, but does not penetrate the cells themselves. Ectomycorrhiza generally associates with woody trees, such as pine, oak, and beech species. These structural differences influence the mechanisms and efficiency of nutrient exchange, although their primary goal remains the same: to enhance the uptake of water and minerals by plants. Together with bacteria, fungi also play a role in the

ammonification process in the nitrogen cycle and the mineralization of phosphorus (Kuyper & Jansa, 2023; Ragonezi & Zavattieri, 2018).

#### **2.4. Protists and Algae: Roles in Food Webs and Carbon Fixation**

Protists, such as protozoa and nematodes, act as predators of other microbes. They prey on populations of bacteria and fungi, and through their waste products, release nutrients back into the environment, thus contributing to nutrient recycling. The role of protists as predators of other microbes demonstrates that biogeochemical cycles involve complex trophic interactions at the micro level, a concept known as the microbial loop. Within the microbial loop, heterotrophic protists (like flagellates and ciliates in soil and water) prey on bacteria and fungi, which have assimilated nutrients from organic matter. Through digestion and excretion, protists convert nutrients bound within microbial biomass into inorganic forms that can be re-accessed by primary producers, such as plants. For instance, it is estimated that protozoa in soil can mineralize up to 30% of the nitrogen immobilized by bacteria, making it available to plants.

They act as regulators of microbial populations, indirectly influencing the rate of nutrient recycling and energy flow within ecosystems. By controlling the population size of bacteria and fungi, protists prevent the "locking up" of nutrients in microbial biomass (the immobilization phenomenon), and instead promote a faster rate of nutrient mineralization. This adds an important layer of ecological complexity to a purely biochemical view of the cycles, demonstrating how predator-prey interactions on a microscopic scale have macroscopic impacts on overall nutrient availability and ecosystem productivity. The presence and activity of protists ensure that vital nutrients like nitrogen and phosphorus continue to flow and remain accessible to plants and other organisms (Neher, 2023).

Algae, along with cyanobacteria, are important photoautotrophs that perform photosynthesis, fixing atmospheric CO<sub>2</sub> into organic biomass within the carbon cycle, especially in aquatic environments. In the oceans, they are major primary producers, forming the base of food webs that support almost all marine life and are responsible for a significant portion of global carbon fixation. Through photosynthesis, they not only convert CO<sub>2</sub> into sugars and other organic compounds but also release oxygen (O<sub>2</sub>) into the atmosphere. This makes them key players in reducing atmospheric CO<sub>2</sub> and acting as significant carbon sinks, helping to mitigate climate change.

In addition to their aquatic roles, some algae and cyanobacteria are also found in terrestrial environments, such as on soil surfaces. There, they can help aerate the soil through several mechanisms. Their colonies, especially cyanobacteria forming biocrusts (biological soil crusts), can produce extracellular polymeric substances (EPS) that bind soil particles, increasing soil aggregation. This process improves soil structure, enhances water infiltration, and reduces erosion. Some cyanobacteria also possess the unique ability to perform atmospheric nitrogen fixation, converting N<sub>2</sub> into forms usable by plants, which significantly increases soil fertility, particularly in desert ecosystems or degraded lands. Thus, they not only support the carbon and oxygen cycles but also directly influence soil health and fertility (Forchhammer & Selim, 2020; Lucius & Hagemann, 2024). Major microbial groups and their contributions to biogeochemical cycles, presented in table 1.

**Table 1.** Major Microbial Groups and Their Biogeochemical Contributions

Microbial Group	Primary Role & Metabolic Diversity	Key Contributions to Biogeochemical Cycles & Examples	Ecological Significance
<b>Bacteria</b>	Indispensable drivers, ubiquitous presence in all environments, vast metabolic diversity.	<b>Nitrogen Cycle:</b> Nitrogen fixation (symbiotic: <i>Rhizobium</i> , free-living: <i>Azotobacter</i> , Cyanobacteria), nitrification ( <i>Nitrosomonas</i> , <i>Nitrobacter</i> ), denitrification ( <i>Pseudomonas</i> , <i>Bacillus</i> ), ammonification. <b>Carbon Cycle:</b> Primary decomposers (release CO <sub>2</sub> ), primary producers (chemoautotrophs, Cyanobacteria), methanogenic and methanotrophic. <b>Phosphorus Cycle:</b> Mineralization of organic P, solubilization of inorganic P, immobilization. <b>Sulfur Cycle:</b> Sulfate reduction, sulfur oxidation (photoautotrophic/chemoautotrophic).	Biological catalysts for element transformation, ensuring efficient nutrient recycling, sustaining productivity, influencing atmospheric composition. Functional redundancy enhances ecosystem resilience.
<b>Archaea</b>	Thrive in extreme environments; major contributors to greenhouse gas cycles.	<b>Carbon Cycle:</b> Primary producers of methane (CH <sub>4</sub> ) via methanogenesis ( <i>Methanococcus</i> , <i>Methanobacteria</i> , <i>Methanosarcina</i> ); perform anaerobic methane oxidation (AOM). Involved in degradation of complex organic matter in sediments/permafrost. <b>Nitrogen Cycle:</b> Ammonia oxidation ( <i>Thaumarchaeota</i> ), especially in oceans.	Crucial for greenhouse gas cycles, especially CH <sub>4</sub> . Adaptability to extreme/changing conditions highlights their importance for climate modeling.
<b>Fungi</b>	Critically important decomposers of complex organic matter; form crucial symbiotic associations.	<b>Decomposition:</b> Break down recalcitrant materials (lignin, cellulose) using extracellular enzymes (laccases, cellulases), releasing CO <sub>2</sub> and inorganic nutrients. Complements bacterial decomposition. <b>Nutrient Acquisition:</b> Mycorrhizal associations (Endomycorrhiza/AM and Ectomycorrhiza) with plant roots enhance uptake of P, N, water. Also involved in ammonification and P mineralization.	Ensure thorough decomposition, increase nutrient availability, extend plant nutrient access, and contribute to soil fertility.
<b>Protists</b>	Predators of other microbes; key players in the microbial loop.	<b>Nutrient Recycling:</b> Consume bacteria and fungi, remineralizing nutrients (e.g., up to 30% of N immobilized by bacteria), making them available to primary producers. <b>Population Regulators:</b> Control microbial populations, preventing nutrient immobilization and promoting faster mineralization.	Add ecological complexity to biogeochemical cycles via trophic interactions, ensuring continuous flow of vital nutrients.
<b>Algae</b>	Important photoautotrophs.	<b>Carbon Cycle:</b> Major primary producers in aquatic environments (oceans), fixing atmospheric CO <sub>2</sub> into organic biomass and releasing O <sub>2</sub> . Significant carbon sinks. <b>Soil Health (with Cyanobacteria):</b> Aerate soil, produce EPS for aggregation, improve water	Base of aquatic food webs, major global carbon fixers, contribute to atmospheric O <sub>2</sub> and soil health/fertility.

infiltration, reduce erosion. Some Cyanobacteria fix atmospheric nitrogen, increasing soil fertility.

### 3. Role of Microorganisms in Specific Biogeochemical Cycles

All essential elements for life—carbon, nitrogen, phosphorus, and sulfur—share a common origin: stellar nucleosynthesis, subsequently dispersed by supernovae to form earth (Arcones & Thielemann, 2023). On our planet, these elements reside in specific reservoirs like the atmosphere, oceans, lithosphere, and biosphere, before being continuously recycled through biogeochemical cycles. Carbon is abundant in oceans and the lithosphere; nitrogen dominates the atmosphere as N<sub>2</sub>; phosphorus is primarily found in the lithosphere; and sulfur is present in the lithosphere, atmosphere, and hydrosphere. The dynamic exchange between these reservoirs, driven by various geological, physical, and chemical processes, forms the core of each element's cycle (Horne & Goldblatt, 2024).

Microorganisms—including bacteria, archaea, fungi, and protists—act as indispensable primary drivers in every one of these cycles. They mediate nearly all nitrogen transformations (such as fixation, nitrification, and denitrification), serve as key decomposers and carbon fixers, facilitate phosphorus mineralization, and dominate sulfur transformations. Without this microbial activity, the nutrient cycles essential for supporting life on Earth simply couldn't operate efficiently.

#### 3.1. Carbon Cycle

Carbon is the basic building block of life and is continuously recycled between the biosphere and the non-living environment. Microorganisms play a central role in every stage of this cycle.

**Carbon Fixation (Photosynthesis):** A crucial step in the carbon cycle is the fixation of atmospheric CO<sub>2</sub> and its assimilation into organic molecules. Autotrophic organisms, including plants, algae, some bacteria, and some archaea, are capable of converting CO<sub>2</sub> into organic molecules through photosynthesis. Cyanobacteria and certain soil bacteria have the ability to perform photosynthesis, where H<sub>2</sub>S and other reduced compounds serve as electron donors to reduce CO<sub>2</sub>. The most prominent carbon fixation pathways include the Calvin-Benson-Bassham (CBB) cycle and the 3-hydroxypropionate bicycle. In addition to photoautotrophs that utilize light energy, chemoautotrophs that oxidize compounds like ammonia, hydrogen, and sulfur also contribute to carbon fixation. Bacterial photosynthesis reactions can be described as:  $\text{CO}_2 + 2\text{H}_2\text{A} \xrightarrow{\text{Light, bacteriochlorophyll}} (\text{CH}_2\text{O})_x + \text{H}_2\text{O} + 2\text{A}$  (Li et al, 2024).

**Decomposition:** Decomposition is a biological process involving the physical breakdown and biochemical transformation of complex organic molecules from dead materials into simpler organic and inorganic molecules. The primary function of soil microorganisms as decomposers is to prevent the permanent sequestration of carbon in complex organic matter, thereby releasing macro and micronutrients that can be taken up by plants. Bacteria and fungi are highly effective at breaking down organic matter, while actinomycetes are responsible for the degradation of recalcitrant residues, such as lignin and chitin. This process releases CO<sub>2</sub>, small molecules, organic molecules, and nitrogen components into the soil (Soong et al, 2020).



Decomposition can be aerobic or anaerobic:

- **Aerobic Decomposition:** Most heterotrophic microbes utilize soil organic compounds aerobically for energy metabolism and as a carbon source. This process produces CO<sub>2</sub>, H<sub>2</sub>O, and energy through mineralization, assimilation, and respiration. Mineralization is the process where organic matter breaks down to release simpler inorganic compounds like CO<sub>2</sub>, NH<sub>4</sub><sup>+</sup>, CH<sub>4</sub>, and H<sub>2</sub>.
- **Anaerobic Decomposition (Methanogenesis):** Anaerobic degradation of carbonaceous material is a collaborative effort involving many bacteria, producing H<sub>2</sub>, CH<sub>4</sub>, alcohols, and organic acids.<sup>4</sup> This process occurs under anoxic conditions, such as deep compacted sludge. Methanogenic bacteria, such as *Methanococcus*, *Methanobacteria*, and *Methanosarcina*, convert acetic acid into CH<sub>4</sub>, CO<sub>2</sub>, and H<sub>2</sub>O, or obtain CH<sub>4</sub> from hydrogen oxidation and CO<sub>2</sub> reduction. Methane (CH<sub>4</sub>) is a very potent greenhouse gas (Huang et al, 2020).

**Methane Oxidation (Methanotrophy):** To complete the recycling pattern, another group of methane bacteria called methanotrophs (*Pseudomonas* and *Methylomonas*) are capable of re-oxidizing the released CH<sub>4</sub> into CO<sub>2</sub>, producing water and energy (Guerrero-Cruz et al, 2021)

**Respiration:** Consumers, including microorganisms, obtain the carbon they need either directly by eating plants or indirectly by eating animals that have eaten plants. This carbon is then used to build their cellular material or released into the atmosphere through respiration. In respiration, most carbohydrates are oxidized to produce CO<sub>2</sub>, H<sub>2</sub>O, and energy, as shown by the equation: C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + 6O<sub>2</sub> -- (Respiration)--> 6CO<sub>2</sub> + 6H<sub>2</sub>O + energy (Tang et al, 2022).

These microbial processes (fixation, decomposition, respiration, methanogenesis, methanotrophy) maintain the dynamic balance of carbon on Earth. Changes in microbial activity, for example due to increased temperatures or land-use changes, can accelerate decomposition, leading to increased levels of CO<sub>2</sub> and CH<sub>4</sub> in the atmosphere. This directly affects climate regulation. The crucial role of microbes in producing and reducing greenhouse gases indicates that understanding and potentially manipulating microbial activity is essential for climate change mitigation efforts. The balance between methanogenesis and methanotrophy, for example, is key in the global methane budget.

### 3.2. Nitrogen Cycle

Nitrogen is a crucial component of nucleic acids and proteins, and it is vital for agriculture. However, incorporating nitrogen into living systems is challenging because plants and phytoplankton cannot directly use atmospheric nitrogen (N<sub>2</sub>), which makes up about 78% of the atmosphere. Microorganisms play a key role in the nitrogen cycle, facilitating the conversion of nitrogen between different forms.

**Nitrogen Fixation:** Nitrogen enters the living world primarily through nitrogen fixation, a process carried out by free-living and symbiotic nitrogen-fixing bacteria. These bacteria convert atmospheric nitrogen gas (N<sub>2</sub>) into macromolecules. Cyanobacteria are important nitrogen fixers found in most aquatic ecosystems exposed to sunlight. Rhizobium bacteria live symbiotically within the root nodules of legumes (such as peas, beans, and peanuts) and provide these plants with the organic nitrogen

they require. Free-living bacteria, such as *Azotobacter*, also contribute significantly to nitrogen fixation (Zhang et al, 2020).

The nitrogen fixation process is highly complex and energy-intensive, mediated by a unique enzyme complex called nitrogenase. Molecularly, the nitrogenase enzyme is a metalloenzyme highly sensitive to oxygen, and it consists of two main proteins. The first is the MoFe-protein component (dinitrogenase); this larger part contains a molybdenum-iron cofactor (MoFe-cofactor) group, which serves as the active site for  $N_2$  binding and reduction. The second is the Fe-protein component (dinitrogenase reductase), a smaller part that functions as an electron donor, transferring electrons one by one to the MoFe-protein using energy from ATP hydrolysis. The overall reaction catalyzed by nitrogenase is  $N_2 + 8H^+ + 8e^- + 16ATP \rightarrow 2NH_3 + H_2 + 16ADP + 16Pi$ .

Because nitrogenase is highly susceptible to oxygen damage, nitrogen-fixing bacteria have developed various protective mechanisms. For example, *Rhizobium* bacteria within legume root nodules create an anaerobic (oxygen-free) environment by producing the oxygen-binding protein leghemoglobin, similar to hemoglobin in blood. This type of molecular adaptation ensures that this vital enzyme can function to convert abundant atmospheric nitrogen into a form assimilable by the biosphere (Einsle & Rees, 2020).

**Ammonification:** This is the first step in the conversion of organic nitrogen back into nitrogen gas in terrestrial systems. During ammonification, certain bacteria and fungi convert nitrogenous waste from living animals or the remains of dead animals into ammonium ( $NH_4^+$ ).

**Nitrification:** This process involves two sub-steps: First, nitrifying bacteria, such as *Nitrosomonas*, convert ammonium ( $NH_4^+$ ) into nitrites ( $NO_2^-$ ). Subsequently, similar organisms convert nitrites ( $NO_2^-$ ) into nitrates ( $NO_3^-$ ).

**Denitrification:** In this final step, bacteria like *Pseudomonas* and *Clostridium* convert nitrates ( $NO_3^-$ ) back into nitrogen gas ( $N_2$ ), allowing it to re-enter the atmosphere. A similar cycle of ammonification, nitrification, and denitrification is also performed by marine bacteria in the marine nitrogen cycle (Zhang et al, 2020).

Microbial transformations in the nitrogen cycle are essential for making nitrogen bioavailable to plants and other organisms. However, imbalances in this cycle, for example due to fertilizer runoff, can lead to excess nitrogen, which in turn triggers eutrophication in aquatic ecosystems. Eutrophication is a process where nutrient runoff causes the excessive growth of microorganisms, which then depletes dissolved oxygen levels and kills ecosystem fauna. This highlights the fragile balance and serious environmental consequences of disrupting microbial processes.

### 3.3. Phosphorus Cycle

Phosphorus is an essential element for all living organisms, but most phosphorus compounds found in nature are in solid and insoluble forms, making them not readily available to plants and animals. Microorganisms play a crucial role in making this phosphorus available.

**Phosphorus Solubilization:** Microorganisms such as *Actinomycetes*, *Pseudomonas*, *Bacillus*, *Aspergillus*, and *Penicillium* are involved in the solubilization of phosphorus in the soil. These microorganisms produce acids that help dissolve

inorganic phosphorus from rocks and sedimentary deposits, thereby making it available in a form that can be absorbed by plants. Symbiotic microbes can also help provide phosphorus to plants (Turchyn et. al, 2018).

Recent discoveries indicate that iron oxides, naturally occurring minerals found in soils and sediments, can also catalyze the cleavage of phosphorus from organic matter at a rate similar to soil enzymes. This suggests a more complex interaction between biological and geological processes than previously thought, where the phosphorus cycle is not solely driven by microbes but also involves significant abiotic contributions. The phenomenon of "missing phosphorus" occurs where inorganic phosphorus, which should be available for life, becomes inaccessible or "locked up" in the environment. This is because phosphate ions (the form of phosphorus absorbable by plants and microbes) strongly attach (adsorb) to the surface of certain minerals, especially abundant iron oxides found in soil and sediments. This process effectively immobilizes phosphorus, making it unavailable to organisms even if the total amount of phosphorus in the soil is high. The "missing phosphorus" phenomenon presents a significant challenge in understanding the phosphorus cycle and nutrient management, as the scarcity of bioavailable phosphorus can limit ecosystem growth and productivity (Moris, 2025).

**Phosphorus Mineralization:** Phosphorus present in living organisms can be returned to reservoirs in the lithosphere through the action of decomposing microorganisms on dead plants and animals. In this step, organic forms of phosphorus are converted into inorganic forms by the process of mineralization. Various decomposing or saprophytic microorganisms such as fungi and bacteria are involved in this step to balance phosphorus concentrations in the ecosystem (Duhamel, 2025).

### 3.4. Sulfur Cycle

Sulfur has many redox states and is a major metabolite in suboxic and anaerobic environments, including marine and marginal marine sediments, the water column of oxygen minimum zones, salt marshes, and oil wells (Turchyn et. al, 2018). Sulfur is an essential component of many proteins.

**Sulfur Oxidation:** Chemolithotrophic sulfur-oxidizing prokaryotes (bacteria like *Beggiatoa*, *Thiobacillus*, *Acidithiobacillus*, *Sulfurimonas*; archaea like *Sulfolobales*) oxidize reduced sulfur compounds ( $H_2S$ ,  $S^0$ ,  $S_2O^{32-}$ ,  $SO^{32-}$ ) to obtain energy, often coupled with the reduction of oxygen or nitrate as terminal electron acceptors. This process contributes to the degradation of organic matter, linking the sulfur and carbon cycles, and controlling methane production and consumption, which is crucial for the long-term  $O_2$  budget in the atmosphere (Turchyn et. al, 2018). Electrogenic sulfur oxidation by cable bacteria is a newly discovered mechanism, involving the formation of multicellular bridges that connect sulfide oxidation in anoxic sediment layers with oxygen or nitrate reduction in oxic surface sediments.

**Sulfur Reduction:** Sulfate-reducing bacteria (*Desulfovibrio desulfuricans*, *Desulfovibrio vulgaris*) reduce oxidized sulfur compounds (sulfate, sulfite) to hydrogen sulfide ( $H_2S$ ) in anaerobic environments, often to obtain energy (dissimilatory sulfate reduction). Some bacteria can also reduce small amounts of sulfate for biosynthesis (assimilatory sulfate reduction) (Geerlings et al, 2019).

The crucial role of microbes in sulfur transformation, especially in suboxic and anaerobic environments like marine sediments and hydrothermal vents, is highly prominent. Hydrothermal vent systems on the seafloor play a pivotal role in the sulfur cycle, acting as unique geochemical and biological hotspots. These vents release hot, mineral-rich fluids, including hydrogen sulfide ( $H_2S$ ), which originates from the interaction of seawater with the hot earth's crust. This  $H_2S$  becomes a vital primary sulfur source in the dark deep-sea environment (Zhou et al, 2022). Chemosynthetic microorganisms, primarily bacteria and archaea, dominate sulfur transformations here; they oxidize  $H_2S$  into sulfate or elemental sulfur as an energy source to fix carbon, forming the base of a unique, lightless food web. In addition to oxidation, sulfate-reducing bacteria can also convert sulfate back into  $H_2S$  in anoxic zones. The released sulfur can also react with dissolved metals to form solid sulfide minerals, contributing to the structure of hydrothermal vent chimneys. Thus, hydrothermal vent systems significantly impact the global oceanic sulfur budget and are a key example of how geological and biological processes closely interact in the biogeochemical cycle of this important element (Turchyn et. al, 2018).

#### 4. Environmental Factors Influencing Microbial Activity

Microbial activity is highly sensitive to environmental conditions, which directly affect the rate and efficiency of biogeochemical cycles.

**Temperature:** Temperature is an important environmental factor influencing microbial activity and enzyme kinetics. Higher temperatures generally accelerate decomposition. Microbes are categorized based on their optimum temperature range: mesophiles (25-40°C, including most pathogens), psychrophiles (<20°C), and thermophiles (55-80°C). Extreme temperatures can lead to microbial inactivation; high temperatures cause desiccation and protein denaturation, while low temperatures can lead to ice crystal formation. Increased temperatures can accelerate carbon cycling mediated by archaea (Fanin et al, 2022).

**pH:** pH directly affects microbial enzymatic activity, cellular structural integrity, and nutrient availability. Each microbial enzyme has its own optimal pH range, and pH changes outside this range can damage its structure, reducing its catalytic activity, which in turn impacts the rate of element transformation. Extreme pH can also damage microbial cells, inhibiting growth or causing death; microbes are categorized as neutrophiles (neutral pH), acidophiles (low pH), or alkaliphiles (high pH) based on their preference. Furthermore, pH influences the chemical form and solubility of nutrients in the environment, affecting the availability of essential elements like phosphorus. As a strong selective factor, pH also shapes microbial community composition, which ultimately influences dominant biogeochemical pathways (Wei et al, 2022).

**Oxygen Availability (Redox Conditions):** Oxygen availability determines the dominant decomposition pathways and metabolic strategies. Aerobic respiration occurs in well-aerated soils, while anaerobic processes such as fermentation, methanogenesis, and dissimilatory sulfate reduction dominate in waterlogged or anoxic environments. Based on oxygen requirements, bacteria are classified into obligate aerobes (*Thiobacillus* spp., which convert sulfides to sulfates in oxygen-rich environments), facultative anaerobes (*Pseudomonas* spp., involved in denitrification),

obligate anaerobes (*Methanobacterium* spp., which produce methane in anoxic swamps), and microaerophiles (*Gallionella ferruginea*, a neutrophilic iron-oxidizing bacterium found in oxic-anoxic transition zones) (Canfield & Kraft 2022).

**Moisture:** Water is an essential component for microbial growth. As a universal solvent, water enables enzymatic and metabolic reactions, facilitates nutrient transport into microbial cells, and allows for waste removal. Optimal moisture levels ensure microbial enzymes function efficiently in breaking down complex substrates and transforming elements like carbon, nitrogen, phosphorus, and sulfur. Water scarcity (desiccation) can halt active microbial growth and biogeochemical activity, even causing cell damage and death, while excessive moisture can create anoxic conditions that alter decomposition pathways. Furthermore, humidity also impacts microbial community structure; for instance, fungi tend to be more dominant in dry conditions compared to bacteria (Bogati, et al, 2025)

**Nutrient Availability (Substrate Quality):** The chemical composition of organic matter affects the rate of decomposition; labile compounds (like sugars) decompose rapidly, while recalcitrant materials (like lignin) decompose more slowly.<sup>4</sup> Nutrient availability also determines dominant metabolic strategies and competitive dynamics within microbial communities. Soil nutrient deficiency due to leaching can significantly affect microbial activity.

In addition to these primary factors, microbial growth is also influenced by CO<sub>2</sub> concentration, light, osmotic effects, and mechanical and sonic stress (Campbell et al, 2015). Summary of the environmental factors influencing microbial activity in biogeochemical cycles, presented in a table 2.

**Table 2.** Impact of Environmental Conditions on Microbial Activity and Biogeochemical Cycle

Environmental Factor	Influence on Microbial Activity & Biogeochemical Cycles	Key Aspects & Examples
<b>Temperature</b>	Affects reaction rates and enzyme activity; influences decomposition speed.	<b>Mesophiles</b> (25-40°C), <b>Psychrophiles</b> (<20°C), <b>Thermophiles</b> (55-80°C). Extremes cause inactivation (desiccation, denaturation, ice crystals). Higher temps accelerate carbon cycling.
<b>pH</b>	Directly impacts enzyme function, cell integrity, and nutrient availability.	Each enzyme has optimal pH. Extreme pH damages cells. Microbes classified as <b>Neutrophiles</b> , <b>Acidophiles</b> , <b>Alkaliphiles</b> . Affects nutrient solubility (e.g., phosphorus) and microbial community composition.
<b>Oxygen Availability (Redox Conditions)</b>	Determines dominant metabolic pathways and decomposition processes.	<b>Aerobic respiration</b> (oxygen present) vs. <b>Anaerobic processes</b> (fermentation, methanogenesis, sulfate reduction) in anoxic environments. Microbes classified as: <b>Obligate Aerobes</b> ( <i>Thiobacillus</i> ), <b>Facultative Anaerobes</b> ( <i>Pseudomonas</i> ), <b>Obligate Anaerobes</b> ( <i>Methanobacterium</i> ), <b>Microaerophiles</b> ( <i>Gallionella</i> ).
<b>Moisture</b>	Essential for microbial growth, metabolic reactions, and nutrient transport.	Optimal levels ensure efficient enzyme function. Water scarcity (desiccation) halts activity and damages cells. Excess moisture leads to anoxic conditions, altering decomposition pathways. Influences microbial community structure (e.g., fungi in dry soils).

<b>Nutrient Availability (Substrate Quality)</b>	Influences decomposition rates and microbial metabolic strategies.	<b>Labile compounds</b> (sugars) decompose quickly; <b>recalcitrant materials</b> (lignin) decompose slowly. Affects competitive dynamics. Nutrient deficiency (e.g., from leaching) can impair microbial activity.
<b>Other Factors</b>	CO <sub>2</sub> concentration, light, osmotic effects, mechanical/sonic stress.	Can also influence microbial growth and overall viability.

## 5. Ecological and Environmental Implications

### 5.1. Global Climate Regulation

Microorganisms have a significant influence on climate change by producing and reducing greenhouse gases (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O, O<sub>3</sub>). Decomposition processes naturally release CO<sub>2</sub> and CH<sub>4</sub> into the atmosphere. Dinitrogen oxide (N<sub>2</sub>O) gas, also a potent greenhouse gas, is produced during nitrification and denitrification processes. Changes in decomposition rates due to climate change, such as increased temperatures and altered precipitation patterns, can increase atmospheric CO<sub>2</sub> levels. Microbes can respond positively or negatively to temperature changes, meaning their activity can change with global temperature fluctuations. Some bacteria, like *Pseudomonas syringae*, can even act as biological ice nucleators at low temperatures, affecting ice crystallization phenomena (Pandey, 2016).

Microbes play a unique and crucial dual role in climate change: they act as both buffers and drivers. As buffers, they are vital in natural climate regulation through processes like carbon fixation, where they absorb CO<sub>2</sub> from the atmosphere and oceans, and methane oxidation, which reduces this potent greenhouse gas in the air. However, microbes are also prone to becoming drivers, as under global warming, higher temperatures can accelerate the decomposition of organic matter, releasing more CO<sub>2</sub> and CH<sub>4</sub> into the atmosphere. Increased temperatures can also trigger the emission of other greenhouse gases like N<sub>2</sub>O from microbial activity. This dual role makes understanding microbial responses to climate change critically important for accurate climate modeling and effective mitigation strategies. The fact that microbial components are often overlooked in current climate models represents a significant gap in our ability to predict Earth's climate future (Tiedje et al, 2022).

### 5.2. Soil Fertility and Sustainable Agriculture

Decomposition and nutrient recycling are crucial for ecosystem function, supporting plant growth and productivity. Microorganisms provide essential nutrients to plants through processes like nitrogen fixation and phosphorus solubilization. Soil microorganisms, including bacteria, actinomycetes, fungi, algae, and protozoa, contribute to soil health and diversity. They help aerate the soil, allowing oxygen and water to penetrate more easily for root development. Mutualistic symbioses, such as between *Rhizobium* bacteria and legumes, or mycorrhizal fungi and plant roots, significantly enhance soil fertility (Tahat et al, 2020).

However, anthropogenic activities like land-use change, excessive nitrogen fertilization, and the use of plastic mulches can alter microbial ecosystems and soil health. Nevertheless, there's significant potential to leverage microbial diversity for sustainable agriculture, particularly through microbiome-based agricultural

innovations. These innovations focus on utilizing and engineering microbial communities to enhance agricultural productivity in an environmentally friendly manner, reducing reliance on chemical fertilizers and pesticides. Examples include the use of biofertilizers (biological fertilizers) involving nitrogen-fixing bacteria (*Azotobacter*) and phosphate-solubilizing bacteria (*Pseudomonas*) to boost nutrient availability. Additionally, biopesticides and biofungicides utilize microbes like *Bacillus thuringiensis* or *Trichoderma* spp. for natural pest and disease control. Endophytic microbes can also be harnessed to improve plant stress tolerance against drought or salinity. Ultimately, through soil microbiome engineering, we can build more resilient and productive microbial ecosystems, becoming a key to a more sustainable global food future (Aloo et al, 2022).

### 5.3. Environmental Purification and Bioremediation

Microorganisms play a vital role in degrading various types of pollutants, maintaining the overall quality and health of the environment—not just soil, but also water and air—through their enzymatic activities. Microbes act as decomposers for dead organisms and various pollutants, effectively "cleaning" different media. In soil, they prevent the accumulation of dead organic matter and contaminants that can seep into groundwater, and are utilized for the bioremediation of oil spills and heavy metals. In aquatic systems like rivers, lakes, oceans, and groundwater, microbes degrade dissolved organic pollutants from waste, clean up marine oil spills, and are used in wastewater treatment plants. Meanwhile, in the air, microorganisms are employed in human-made systems such as biofilters and bioscrubbers to degrade volatile organic compounds (VOCs) and other gaseous pollutants. The remarkable ability to engineer microorganisms for specific tasks, such as breaking down plastics or degrading stubborn chemicals, highlights their immense potential in reducing our environmental footprint and providing sustainable solutions for global pollution challenges (Ayilara et al, 2023).

One well-known example of successful bioremediation is the handling of the Exxon Valdez oil spill in 1989 in Alaska. Here, the addition of nutrients (biostimulation) significantly accelerated the degradation of hydrocarbons by native bacteria, reducing oil concentrations by up to 70% within a year. Another case is the Deepwater Horizon oil spill in 2010 in the Gulf of Mexico, where naturally occurring oil-eating bacteria like *Alcanivorax borkumensis* played a crucial role. Bioremediation has also been successfully applied to clean up heavy metal-contaminated soil, such as in Bangladesh and India, where arsenic-resistant bacteria convert toxic inorganic arsenic into less harmful organic forms. These successful case studies demonstrate that microorganisms, both natural and engineered, are powerful and sustainable tools in environmental pollution mitigation efforts (Alabssawy & Hashem, 2024; Dilpazeer et al, 2023)).

### 5.4. Impact of Anthropogenic Activities

Human activities are profoundly impacting global carbon and nutrient biogeochemical cycles. Land-use changes like deforestation and agriculture alter organic matter input and decomposition, potentially depleting soil organic carbon and fertility. Pollution introduces novel substrates, selecting for tolerant microbial species,

while nitrogen fertilization and atmospheric deposition significantly reshape microbial ecosystems. A notable interaction involves microplastic contamination, which can reduce soil water retention and plant biomass, and negatively affect soil protist communities. Interestingly, nitrogen fertilization can mitigate these microplastic-induced negative effects. This mitigation occurs because nitrogen boosts microbial activity and populations, enhancing their ability to cope with stress or form biofilms on microplastics. It can also alter microbial community composition, favoring more tolerant species. Additionally, sufficient nitrogen availability improves plant health, making them more resilient to microplastic impacts. Despite these insights from ongoing research, nitrogen fertilization is not a long-term solution to microplastic pollution. The broader impact of human activities on microbial diversity is a growing concern, as habitat destruction, pollution, and climate change threaten these vital communities essential for ecosystem health (Raimi et al, 2021).

## 6. Conclusion

Microorganisms are unseen yet indispensable drivers in global biogeochemical cycles. They are not merely participants, but fundamental ecological architects that facilitate the transformation of essential elements such as carbon, nitrogen, phosphorus, and sulfur, which underpin the sustainability of life on Earth. Their metabolic diversity, encompassing bacteria, archaea, fungi, algae, and protists, allows them to dominate various environments and perform vital functions ranging from atmospheric gas fixation, decomposition of complex organic matter, to environmental purification.

The complex interactions between different microbial groups, including functional division of labor and trophic interactions, ensure the efficiency and resilience of nutrient cycles. However, microbial activity is highly influenced by environmental factors such as temperature, pH, oxygen availability, and moisture. Changes in these factors, whether naturally occurring or due to anthropogenic activities, can significantly alter the rate and direction of biogeochemical cycles.

The crucial role of microorganisms has profound implications for global challenges such as climate change, food security, and environmental pollution. They are dual agents that can buffer against climate change through carbon fixation and methane oxidation, but can also accelerate climate change through increased greenhouse gas emissions due to accelerated decomposition. Therefore, a deeper understanding of microbial ecology and its intricate connections with biogeochemical cycles is paramount.

Future research must decisively focus on unraveling the microbial "dark matter"—that is, the unidentified diversity and functions of microorganisms—and deeply understanding the complex responses of microbial communities to various environmental pressures. This crucial knowledge will serve as an irreplaceable foundation for developing far more effective strategies in environmental management, climate change mitigation, and sustainable agricultural practices.

To achieve this vision, research must be highly interdisciplinary. This means close collaboration among microbiology, ecology, soil science, oceanography, climatology, data science, and genetic engineering. We need to integrate advanced metagenomic, metatranscriptomic, and metaproteomic technologies to uncover the identity and functions of unknown microbes. Concurrently, climate modeling must



explicitly integrate microbial dynamics and responses, moving beyond approaches that have historically overlooked their central role. Furthermore, the development of microbe-based solutions for bioremediation and enhanced agricultural productivity must be balanced with a strong ecological understanding to ensure sustainability and environmental safety. Only with this holistic and integrated approach can we truly comprehend and harness the unseen power of the microbial world to support global ecosystem health and human well-being.

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